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Prelims

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1) a. adiabatic $\Rightarrow \delta Q = 0$

$$dU = \delta W = -P dV$$

$$dU = d(VU(T)) = U(T)dV + VU'(T)dT$$

$$U(T)dV + VU'(T)dT = -\alpha U(T)dV$$

$$VU'(T)dT = -(\alpha+1)U(T)dV$$

$$\frac{dV}{dT} = -\frac{V}{\alpha+1} \frac{U'(T)}{U(T)}$$

$$-dV \frac{\alpha+1}{V} = \frac{U'(T)}{U(T)} dT$$

$$-(\alpha+1) \ln(V) = \ln(U(T)) + C$$

$$\ln(U(T) \cdot V^{\alpha+1}) = \text{const}$$

$\therefore U(T)V^{\alpha+1}$ is constant for an adiabatic process

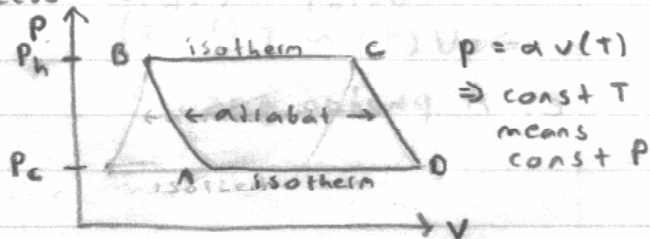
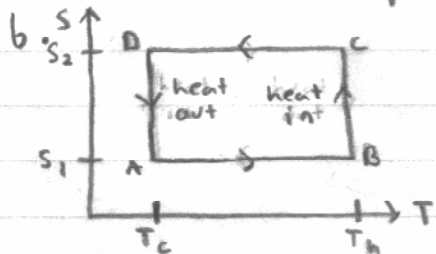
isothermal $\Rightarrow dT = 0$

$$dU = \delta Q + \delta W$$

$$U(T)dV = \delta Q - \alpha U(T)dV$$

$$\delta Q = (\alpha+1)U(T)dV$$

For an isothermal process $Q = (\alpha+1)U(T)\Delta V$



$$Q_{in} = (\alpha+1)U(T_h)(V_c - V_b) \quad U(T)V^{\alpha+1} = C$$

$$Q_{out} = (\alpha+1)U(T_c)(V_d - V_a)$$

$$dU_{cycle} = 0 \Rightarrow W_{one} = Q_{in} - Q_{out}$$

$$W_{one} = (\alpha+1)U(T_h)(V_c - V_b) - (\alpha+1)U(T_c)(V_d - V_a)$$

$$U(T_h) V_B^{\alpha+1} = U(T_c) V_A^{\alpha+1}$$

$$V_A = V_B \left(\frac{U(T_h)}{U(T_c)} \right)^{\frac{1}{\alpha+1}}$$

$$V_D = V_C \left(\frac{U(T_h)}{U(T_c)} \right)^{\frac{1}{\alpha+1}}$$

$$\eta = \frac{W_{\text{done}}}{Q_{\text{in}}}$$

$$= \frac{\alpha+1}{\alpha+1} \left[1 - \frac{U(T_c)}{U(T_h)} \left(\frac{V_D - V_A}{V_C - V_B} \right) \right]$$

$$= \frac{\alpha+1}{\alpha+1} \left[1 - \frac{U(T_c)}{U(T_h)} \left(\frac{U(T_h)}{U(T_c)} \right)^{\frac{1}{\alpha+1}} \right]$$

$$= \frac{\alpha+1}{\alpha+1} \left[1 - \left(\frac{U(T_c)}{U(T_h)} \right)^{\alpha/\alpha+1} \right]$$

$$\eta = 1 - \left(\frac{U(T_c)}{U(T_h)} \right)^{\alpha/\alpha+1}$$

$$c. \eta = \frac{W_{\text{done}}}{Q_{\text{in}}}$$

$$= \frac{Q_{\text{in}} - Q_{\text{out}}}{Q_{\text{in}}} \quad \begin{aligned} Q_{\text{in}} &= T_h \Delta S \\ Q_{\text{out}} &= T_c \Delta S \end{aligned}$$

$$\eta = 1 - \frac{Q_{\text{out}}}{Q_{\text{in}}}$$

Since $\Delta S = S_2 - S_1$ is the same for both isotherms:

$$\eta = 1 - \frac{T_c}{T_h} \cdot \frac{\Delta S}{\Delta S}$$

$$\eta = 1 - \frac{T_c}{T_h}$$

$$d. i. \left(\frac{U(T_c)}{U(T_h)} \right)^{\alpha/\alpha+1} = \frac{T_c}{T_h}$$

$$\frac{U(T_c)}{U(T_h)} = \left(\frac{T_c}{T_h} \right)^{\frac{\alpha+1}{\alpha}}$$

$$\Rightarrow U(T) \sim T^{(1 + \frac{1}{\alpha})}$$

e. A photon gas