

# 1 Thermodynamics

## 1.1 Problem 1

(a)

This is a chemical potential problem. We must know the chemical potential of an ideal gas, or at least know how to compute it relatively quickly. It is:

$$\mu = kT \ln \left( \frac{\rho \pi \hbar^3 \sqrt{2mkT\pi}}{m^2(kT)^2} \right)$$

where  $\rho$  is the number density  $N/V$ . So in our case, taking into account the binding energy of hydrogen, we have:

$$-\epsilon + \mu_H(\epsilon = 0) = \mu_P + \mu_e$$

In this first part we are taking all the densities to be the same. If we assume  $m_H = m_P$  (in reality  $m_H = m_P * 1.0005$ ), we can cancel the hydrogen and proton terms:

$$\begin{aligned} kT \ln \left( \frac{\rho \pi \hbar^3 \sqrt{2m_e kT \pi}}{m_e^2 (kT)^2} \right) &= -\epsilon \\ \rho &= \frac{e^{-\epsilon/kT} m_e^2 (kT)^2}{\pi \hbar^3 \sqrt{2m_e kT \pi}} \approx 1.92 \cdot 10^{-17} m^{-3} \end{aligned} \quad (1)$$

(b)

It's hard to interpret what they mean by "at this density". Do they mean "three times  $\rho$ "? Or the density of hydrogen atoms remains  $\rho$  while the others decrease? I don't know, but I'll take it to mean the latter, so that, after the reduction in temperature:

$$\begin{aligned} \rho_H &= \rho \\ \rho_P &= \rho_e = \frac{\rho}{9} \end{aligned}$$

We plug these into the equation relating the chemical potentials:

$$-\epsilon + kT \ln \left( \frac{\rho \pi \hbar^3 \sqrt{2m_H kT \pi}}{m_H^2 (kT)^2} \right) = kT \ln \left( \frac{\rho \pi \hbar^3 \sqrt{2m_P kT \pi}}{9m_P^2 (kT)^2} \right) + kT \ln \left( \frac{\rho \pi \hbar^3 \sqrt{2m_e kT \pi}}{9m_e^2 (kT)^2} \right)$$

Applying the approximation  $m_H \approx m_P$ , we can cancel most of the proton and hydrogen terms. Then we plug in  $\rho$  from part (a), with  $T_i$  meaning the initial temperature:

$$\begin{aligned} -\epsilon &= -kT \ln 9 + kT \ln \left( \frac{e^{-\epsilon/kT_i} m_e^2 (kT_i)^2 \pi \hbar^3 \sqrt{2m_e kT \pi}}{\pi \hbar^3 \sqrt{2m_e kT_i \pi} 9m_e^2 (kT)^2} \right) \\ -\epsilon &= -kT \ln 9 + kT \ln \left( \frac{e^{-\epsilon/kT_i} T_i^{3/2}}{9T^{3/2}} \right) \end{aligned}$$

$$kT \ln(e^{-\epsilon/kT}) = kT \ln\left(\frac{e^{-\epsilon/kT_i} T_i^{3/2}}{81T^{3/2}}\right)$$

$$e^{\epsilon/kT_i - \epsilon/kT} = \frac{T_i^{3/2}}{81T^{3/2}}$$

After staring at this equation for a long time, I realized the only way it can be satisfied is if  $T$  and  $T_i$  are close to each other. You can then write  $T = T_i + \delta$ , and solve for  $\delta$ . You end up getting:

$$kT \approx 0.955kT_i = 0.130eV \quad (2)$$